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Key indicators

Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(C-C) = 0.009$ Å
 R factor = 0.039
 wR factor = 0.094
Data-to-parameter ratio = 11.6

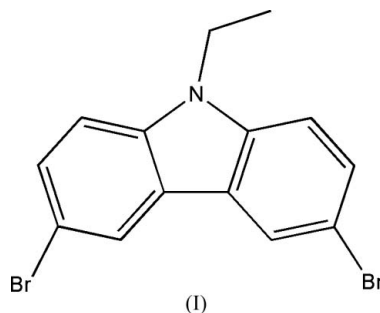
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

3,6-Dibromo-9-ethyl-9H-carbazole

The title compound, $C_{14}H_{11}Br_2N$, was synthesized by *N*-alkylation of bromoethane with 3,6-dibromo-9H-carbazole. The carbazole ring system is essentially planar and makes a dihedral angle of $94.7(3)^\circ$ with the plane formed by the ethyl C and carbazole N atoms.

Comment

Carbazole derivatives substituted by *N*-alkylation are valuable for anticancer research (Buu-Hoï & Royer, 1950; Caulfield *et al.*, 2002; Morton *et al.*, 1983). In view of the increased toxicity to liver tissue developed in organic molecules by halogenation, emphasis has been placed on the synthesis of carbazole derivatives bearing halogen atoms (Buu-Hoï & Royer, 1951; Richard *et al.*, 2004). The title carbazole compound, (I), has recently been synthesized by *N*-alkylation of bromoethane with 3,6-dibromo-9H-carbazole. We present here its X-ray crystal structure.



The molecular structure of (I) is illustrated in Fig. 1. The carbazole ring system is essentially planar and the ethyl group is out of the carbazole plane. The dihedral angle between the carbazole plane and the plane formed by the ethyl C and carbazole N atoms is $94.7(3)^\circ$.

Experimental

The title compound was prepared according to the procedure of Duan *et al.* (2005). 3,6-Dibromocarbazole (6.5 g, 20 mmol) was added to a dimethylformamide (DMF) solution (50 ml) of KOH (7.0 g) and the mixture was stirred for 40 min. A DMF solution (50 ml) of bromoethane (3.3 g, 30 mmol) was added dropwise with stirring. The resulting mixture was stirred at room temperature for 10 h and then poured into water (500 ml), yielding a white precipitate. The solid product was filtered, and washed with cold water. Crystallization from EtOH gave fine crystals of (I) (yield: 6.50 g, 92%; m.p. 412–414 K). 1H NMR ($CDCl_3$): δ 1.33 (*t*, $J = 7.1$ Hz), 4.17 (*q*, $J = 7.1$ Hz, 2H), 7.16 (*d*, $J = 8.5$ Hz, 2H), 7.48 (*dd*, $J_1 = 8.5$ Hz, $J_2 = 1.8$ Hz, 2H), 8.13 (*d*, $J = 1.8$ Hz, 2H); ^{13}C NMR ($CDCl_3$): δ 13.64, 37.73, 110.05, 111.92, 123.20, 123.48, 128.95, 138.7. Recrystallization from a mixture

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of chloroform (5 ml) and ethanol (5 ml) gave colorless single crystals of (I) after two weeks.

Crystal data

$C_{14}H_{11}Br_2N$
 $M_r = 353.06$
 Monoclinic, $P2_1$
 $a = 4.2588$ (8) Å
 $b = 10.930$ (2) Å
 $c = 13.724$ (3) Å
 $\beta = 95.733$ (3)°
 $V = 635.6$ (2) Å³
 $Z = 2$

$D_x = 1.845$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 1746 reflections
 $\theta = 3.0$ – 25.7 °
 $\mu = 6.35$ mm⁻¹
 $T = 294$ (2) K
 Block, colorless
 $0.28 \times 0.22 \times 0.20$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 2002)
 $T_{\min} = 0.176$, $T_{\max} = 0.281$
 3520 measured reflections

1804 independent reflections
 1470 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.025$
 $\theta_{\text{max}} = 26.3$ °
 $h = -5 \rightarrow 5$
 $k = -8 \rightarrow 13$
 $l = -14 \rightarrow 17$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.094$
 $S = 1.04$
 1804 reflections
 155 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0473P)^2 + 0.263P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.66$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.50$ e Å⁻³
 Absolute structure: Flack (1983),
 444 Friedel Pairs
 Flack parameter: 0.02 (2)

Table 1

Selected geometric parameters (Å).

Br1—C4	1.907 (7)	N1—C7	1.386 (8)
Br2—C10	1.908 (6)	N1—C13	1.453 (8)
N1—C1	1.371 (8)		

H atoms were positioned geometrically, with C—H = 0.93–0.97 Å and refined using a riding model [$U_{\text{iso}}(\text{H}) = 1.2$ or 1.5 times $U_{\text{eq}}(\text{C})$].

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1998); software used to prepare material for publication: SHELXTL.

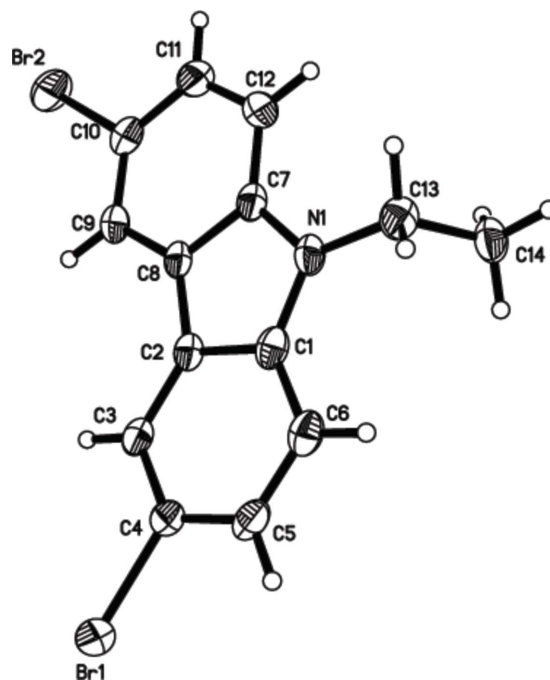


Figure 1

The molecular structure of (I), shown with 30% probability displacement ellipsoids (arbitrary spheres for H atoms).

References

- Bruker (1998). SMART, SAINT and SHELXTL (Version 5). Bruker AXS Inc., Madison, Wisconsin, USA.
 Buu-Hoi, N. P. & Royer R. (1950). *J. Org. Chem.* **15**, 123–130.
 Buu-Hoi, N. P. & Royer R. (1951). *J. Org. Chem.* **16**, 1198–1205.
 Caulfield, T., Cherrier, M. P., Combeau, C. & Mailliet, P. (2002). EP 1253141 A1.
 Duan, X.-M., Han, J., Chen, L.-G., Xu, Y.-J. & Li, Y. (2005). *Fine Chemicals*, **22**, 39–52.
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
 Morton, H. & Charles, T. J. (1983). *US Patent No.* 4379160.
 Richard, W. H., Ho, S. L. & Michael E. R. (2004). *US Patent No.* 2004/0087796 A1.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
 Sheldrick, G. M. (2002). SADABS. University of Göttingen, Germany.